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## Key indicators

Single-crystal X-ray study
$T=298 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.003 \AA$
$R$ factor $=0.027$
$w R$ factor $=0.063$
Data-to-parameter ratio $=15.0$
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.
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# Polymeric disodium ( $\mu$-trans-4,4'-ethylenedipyridyl)bis[ $\mu$-5-sulfoisophthalato(3-)]dicadmate(II) hexahydrate 

The title crystal structure, $\left\{\mathrm{Na}_{2}\left[\mathrm{Cd}_{2}\left(\mathrm{C}_{12} \mathrm{H}_{10} \mathrm{~N}_{2}\right)\left(\mathrm{C}_{8} \mathrm{H}_{3} \mathrm{O}_{7} \mathrm{~S}\right)_{2}\right]\right.$-$\left.6 \mathrm{H}_{2} \mathrm{O}\right\}_{n}$, contains pentagonal-bipyramidal Cd atoms and octahedral Na atoms. The Cd atom is chelated by two carboxyl groups of adjacent 5-sulfoisophthalate(3-) trianions, and it is also coordinated by the pyridyl group of the centrosymmetric heterocycle, the coordinating atoms comprising the pentagonal plane around the metal atom. In the network structure, the Na atoms occupy the voids in the network such that they exist in a six-coordinate geometry.

## Comment

A recent investigation of the reaction between monosodium 5sulfoisophthalate and cadmium nitrate, in the presence of the N -donor ligand 1,4-diazobicyclo[2.2.2]octane, has reported zwitterionic aqua(5-sulfoisophthalato(1-aza-4-azoniabicyclo[2.2.2] octane)cadmate. In the crystal structure, the 5 -sulfoisophthalate trianion interacts with three symmetry-related Cd atoms; the Cd atom itself is coordinated by the monoprotonated N -donor ligand $[\mathrm{Cd} \leftarrow \mathrm{N}=2.437$ (2) $\AA$; Tao et al., 2003]. The analogous reaction, which was carried out in the presence of 4,4'-trans-dipyridylethylene, led to the title Cd complex, (I), having two dative $\mathrm{Cd} \leftarrow$ bonds; the bridging spacer has an inversion center at the mid-point of the central $\mathrm{C}=\mathrm{C}$ bond (Fig. 1 and Table 1).

In the dianion of (I), the Cd atom is chelated by the carboxyl group of one 5 -sulfoisophthalate (3-) trianion [Cd$\mathrm{O}=2.265$ (2) and 2.564 (2) $\AA$, and $\mathrm{O}-\mathrm{Cd}-\mathrm{O}=53.4$ (1) $\left.{ }^{\circ}\right]$ and by that of another trianion $[\mathrm{Cd}-\mathrm{O}=2.306$ (2) and 2.528 (2) $\AA$, and $\left.\mathrm{O}-\mathrm{Cd}-\mathrm{O}=53.6(1)^{\circ}\right]$. The fifth coordination site of the pentagonal plane is occupied by the N atom of the heterocycle $[\mathrm{Cd}-\mathrm{N}=2.285$ (2) $\AA$ ]; the apical sites are occupied by the sulfonyl O atoms $[\mathrm{Cd}-\mathrm{O}=2.342$ (2) and 2.402 (2) $\AA$, and $\left.\mathrm{O}-\mathrm{Cd}-\mathrm{O}=162.9(1)^{\circ}\right]$. The $\mathrm{Cd}-\mathrm{O}$ and $\mathrm{Cd}-\mathrm{N}$ bonds lead to the formation of a network structure; the


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Na atoms occupy the voids in the network such that each exists in a six-coordinate geometry. The network structure is further consolidated by hydrogen-bonding interactions (Table 2).

## Experimental

The hydrothermal synthesis was carried out with cadmium nitrate tetrahydrate ( $0.15 \mathrm{~g}, 0.5 \mathrm{mmol}$ ), the monosodium salt of 5 -sulfoisophthalic acid $(0.13 \mathrm{~g}, 0.5 \mathrm{mmol})$ and $4,4^{\prime}$-dipyridylethylene $(0.09 \mathrm{~g}$, 0.5 mmol ) in the manner used for the preparation of a previous cadmium complex (Tao et al., 2003). A volume of sodium hydroxide solution corresponding to twice the molar equivalent of $\mathrm{Cd}^{2+}$ was used to convert the acid to its trianion.

## Crystal data

$\mathrm{Na}_{2}\left[\mathrm{Cd}_{2}\left(\mathrm{C}_{12} \mathrm{H}_{10} \mathrm{~N}_{2}\right)\left(\mathrm{C}_{8} \mathrm{H}_{3} \mathrm{O}_{7} \mathrm{~S}\right)_{2}\right]$ -

$$
D_{x}=1.963 \mathrm{Mg} \mathrm{~m}^{-3}
$$

$6 \mathrm{H}_{2} \mathrm{O}$
$M_{r}=1047.42$
Monoclinic, $P 2_{\mathrm{d}} / n$
$a=8.9936$ (4) A
$b=16.2164$ (7) $\AA$
$c=12.4217(5) \AA$
$\beta=101.999$ (1) ${ }^{\circ}$
$V=1772.1(1) \AA^{3}$
$Z=2$

## Data collection

Bruker APEX area-detector diffractometer
$\varphi$ and $\omega$ scans
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)
$T_{\text {min }}=0.570, T_{\text {max }}=0.730$
10511 measured reflections

## Refinement

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.027$
$w R\left(F^{2}\right)=0.063$
$S=0.95$
4075 reflections
271 parameters

4075 independent reflections
3510 reflections with $I>2 \sigma(I)$
$R_{\text {int }}=0.022$
$\theta_{\text {max }}=28.3^{\circ}$
$h=-11 \rightarrow 5$
$k=-19 \rightarrow 20$
$l=-12 \rightarrow 16$

H -atom parameters constrained
$w=1 /\left[\sigma^{2}\left(F_{o}^{2}\right)+(0.0365 P)^{2}\right]$
where $P \stackrel{o}{=}\left(F_{o}^{2}+2 F_{c}^{2}\right) / 3$
$(\Delta / \sigma)_{\max }=0.001$
$\Delta \rho_{\text {max }}=0.85 \mathrm{e}_{\mathrm{m}} \mathrm{A}^{-3}$
$\Delta \rho_{\text {min }}=-0.43 \mathrm{e}^{-3}$


Figure 1
ORTEPII (Johnson, 1976) plot of a segment of the title structure, with displacement ellipsoids drawn at the $50 \%$ probability level. [Symmetry codes: (i) $\frac{3}{2}-x, y-\frac{1}{2}, \frac{1}{2}-z$; (ii) $x-\frac{1}{2}, \frac{1}{2}-y, \frac{1}{2}+z$; (iii) $x-1, y, z$.]

Table 1
Selected geometric parameters $\left(\AA{ }^{\circ}{ }^{\circ}\right)$.

| Cd1-O1 | 2.528 (2) | Na1-O1 | 2.291 (2) |
| :---: | :---: | :---: | :---: |
| Cd1-O2 | 2.306 (2) | $\mathrm{Na} 1-\mathrm{O} 3{ }^{\text {i }}$ | 2.686 (2) |
| $\mathrm{Cd} 1-\mathrm{O} 3{ }^{\text {i }}$ | 2.265 (2) | $\mathrm{Na} 1-\mathrm{O} 7^{\text {ii }}$ | 2.421 (2) |
| $\mathrm{Cd} 1-\mathrm{O} 4^{\text {i }}$ | 2.564 (2) | $\mathrm{Na} 1-\mathrm{O} 1 w$ | 2.676 (3) |
| $\mathrm{Cd} 1-\mathrm{O} 5^{\text {ii }}$ | 2.342 (2) | Na1-O2w | 2.384 (3) |
| $\mathrm{Cd} 1-\mathrm{O} 6^{\text {iii }}$ | 2.402 (2) | Na1-O3w | 2.349 (2) |
| Cd1-N1 | 2.285 (2) |  |  |
| $\mathrm{O} 1-\mathrm{Cd} 1-\mathrm{O} 2$ | 53.6 (1) | $\mathrm{O} 5^{\mathrm{ii}}-\mathrm{Cd} 1-\mathrm{O} 6^{\mathrm{iii}}$ | 162.9 (1) |
| $\mathrm{O} 1-\mathrm{Cd} 1-\mathrm{O} 3{ }^{\text {i }}$ | 75.0 (1) | $\mathrm{O} 5^{\mathrm{ii}}-\mathrm{Cd} 1-\mathrm{N} 1$ | 82.2 (1) |
| $\mathrm{O} 1-\mathrm{Cd} 1-\mathrm{O} 4^{\text {i }}$ | 126.6 (1) | $\mathrm{O} 6^{\text {iii }}-\mathrm{Cd} 1-\mathrm{N} 1$ | 80.8 (1) |
| $\mathrm{O} 1-\mathrm{Cd} 1-\mathrm{O} 5^{\text {ii }}$ | 86.2 (1) | $\mathrm{O} 1-\mathrm{Na} 1-\mathrm{O}^{\text {i }}$ | 71.5 (1) |
| $\mathrm{O} 1-\mathrm{Cd} 1-\mathrm{O} 6^{\text {iii }}$ | 109.0 (1) | $\mathrm{O} 1-\mathrm{Na} 1-\mathrm{O} 7^{\mathrm{ii}}$ | 93.3 (1) |
| $\mathrm{O} 1-\mathrm{Cd} 1-\mathrm{N} 1$ | 143.3 (1) | $\mathrm{O} 1-\mathrm{Na} 1-\mathrm{O} 1 w$ | 95.1 (1) |
| $\mathrm{O} 2-\mathrm{Cd} 1-\mathrm{O}^{\text {i }}$ | 122.7 (1) | $\mathrm{O} 1-\mathrm{Na} 1-\mathrm{O} 2 w$ | 113.3 (1) |
| $\mathrm{O} 2-\mathrm{Cd} 1-\mathrm{O} 4{ }^{\text {i }}$ | 170.1 (1) | $\mathrm{O} 1-\mathrm{Na} 1-\mathrm{O} 3 w$ | 163.1 (1) |
| $\mathrm{O} 2-\mathrm{Cd} 1-\mathrm{O} 5^{\text {ii }}$ | 99.9 (1) | $\mathrm{O} 3{ }^{\text {i }}-\mathrm{Na} 1-\mathrm{O} 7^{\text {ii }}$ | 75.6 (1) |
| $\mathrm{O} 2-\mathrm{Cd} 1-\mathrm{O}^{\text {iii }}$ | 83.7 (1) | $\mathrm{O} 3{ }^{\mathrm{i}}-\mathrm{Na} 1-\mathrm{O} 1 w$ | 127.9 (1) |
| $\mathrm{O} 2-\mathrm{Cd} 1-\mathrm{N} 1$ | 94.3 (1) | $\mathrm{O} 3{ }^{\mathrm{i}}-\mathrm{Na} 1-\mathrm{O} 2 w$ | 155.3 (1) |
| $\mathrm{O} 3{ }^{\mathrm{i}}-\mathrm{Cd} 1-\mathrm{O} 4^{\mathrm{i}}$ | 53.4 (1) | $\mathrm{O} 3{ }^{\mathrm{i}}-\mathrm{Na} 1-\mathrm{O} 3 w$ | 95.9 (1) |
| $\mathrm{O} 3{ }^{\mathrm{i}}-\mathrm{Cd} 1-\mathrm{O} 5^{\mathrm{ii}}$ | 100.2 (1) | $\mathrm{O} 77^{\text {ii }}-\mathrm{Na} 1-\mathrm{O} 1 w$ | 156.6 (1) |
| $\mathrm{O} 3^{\text {i }}-\mathrm{Cd} 1-\mathrm{O} 6^{\text {iii }}$ | 91.7 (1) | $\mathrm{O} 7{ }^{\text {ii }}-\mathrm{Na} 1-\mathrm{O} 2 w$ | 79.9 (1) |
| $\mathrm{O} 3^{\mathrm{i}}-\mathrm{Cd} 1-\mathrm{N} 1$ | 141.3 (1) | $\mathrm{O} 77^{\text {ii }}-\mathrm{Na} 1-\mathrm{O} 3 w$ | 94.5 (1) |
| $\mathrm{O} 4^{\mathrm{i}}-\mathrm{Cd} 1-\mathrm{O} 5^{\mathrm{ii}}$ | 90.0 (1) | $\mathrm{O} 1 w-\mathrm{Na} 1-\mathrm{O} 2 w$ | 76.6 (1) |
| $\mathrm{O} 4^{\mathrm{i}}-\mathrm{Cd} 1-\mathrm{O}^{\text {iii }}$ | 87.2 (1) | $\mathrm{O} 1 w-\mathrm{Na} 1-\mathrm{O} 3 w$ | 83.6 (1) |
| $\mathrm{O} 4^{\mathrm{i}}-\mathrm{Cd} 1-\mathrm{N} 1$ | 88.2 (1) | $\mathrm{O} 2 w-\mathrm{Na} 1-\mathrm{O} 3 w$ | 82.9 (1) |

Table 2
Hydrogen-bonding geometry $\left(\AA,{ }^{\circ}\right)$.

| $D-\mathrm{H} \cdots A$ | $D-\mathrm{H}$ | $\mathrm{H} \cdots A$ | $D \cdots A$ | $D-\mathrm{H} \cdots A$ |
| :--- | :--- | :--- | :--- | :--- |
| $\mathrm{O} 1 w-\mathrm{H} 1 w 1 \cdots \mathrm{O} 5$ | 0.86 | 2.26 | $3.094(3)$ | 164 |
| $\mathrm{O} 1 w-\mathrm{H} 1 w 2 \cdots \mathrm{O} 4^{\mathrm{i}}$ | 0.85 | 2.27 | $3.109(4)$ | 170 |
| $\mathrm{O}^{\mathrm{O}} w-\mathrm{H} 2 w 2 \cdots 4^{\mathrm{i}}$ | 0.85 | 2.00 | $2.790(3)$ | 156 |
| $\mathrm{O}^{\mathrm{O}} w-\mathrm{H} 2 w 1 \cdots \mathrm{O}^{\mathrm{iv}} w^{\mathrm{iv}}$ | 0.85 | 2.13 | $2.958(3)$ | 164 |
| $\mathrm{O}^{2} w-\mathrm{H} 3 w 1 \cdots \mathrm{O}^{\mathrm{v}}$ | 0.86 | 1.95 | $2.782(3)$ | 162 |
| ${\mathrm{O} 3 w-\mathrm{H} 3 w 2 \cdots \mathrm{O}^{\mathrm{i}}}$ | 0.85 | 2.10 | $2.887(3)$ | 153 |

Symmetry codes: (i) $\frac{3}{2}-x, y-\frac{1}{2}, \frac{1}{2}-z$; (iv) $\frac{5}{2}-x, y-\frac{1}{2}, \frac{1}{2}-z$; (v) $2-x,-y, 1-z$.
The H atoms bonded to water were located and refined subject to $\mathrm{O}-\mathrm{H}=0.85$ (1) $\AA$ and $\mathrm{H} \cdots \mathrm{H}=1.39$ (1) $\AA$. The C -bound H atoms were positioned geometrically $(\mathrm{C}-\mathrm{H}=0.93 \AA)$ and were included in the refinement in the riding-model approximation. The displacement parameters of all H atoms were set to 1.2 times the equivalent isotropic displacement parameters of their parent atoms.

Data collection: SMART (Bruker, 2001); cell refinement: SAINT (Bruker, 2001); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEPII (Johnson, 1976); software used to prepare material for publication: SHELXL97.

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